



BSI Standards Publication

**Ferronickels — Determination of carbon content — Infrared absorption method after induction furnace combustion**

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## National foreword

This British Standard is the UK implementation of [ISO 7524:2020](#). It supersedes [BS 6783-5:1986](#), which is withdrawn.

Attention is drawn to the fact that BSI is not a member of ISO/TC 155 and the UK did not participate in the development of the standard at ISO level.

The UK participation in its preparation was entrusted to Technical Committee I/-, Miscellaneous standards materials and chemicals.

A list of organizations represented on this committee can be obtained on request to its secretary.

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# INTERNATIONAL STANDARD

# ISO 7524

Second edition  
2020-03-04

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## **Ferronickels — Determination of carbon content — Infrared absorption method after induction furnace combustion**

*Ferronickels — Détermination de la teneur en carbone — Méthode  
par absorption dans l'infrarouge après combustion dans un  
four à induction*



Reference number  
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## Foreword

ISO (the International Organization for Standardization) is a worldwide federation of national standards bodies (ISO member bodies). The work of preparing International Standards is normally carried out through ISO technical committees. Each member body interested in a subject for which a technical committee has been established has the right to be represented on that committee. International organizations, governmental and non-governmental, in liaison with ISO, also take part in the work. ISO collaborates closely with the International Electrotechnical Commission (IEC) on all matters of electrotechnical standardization.

The procedures used to develop this document and those intended for its further maintenance are described in the ISO/IEC Directives, Part 1. In particular, the different approval criteria needed for the different types of ISO documents should be noted. This document was drafted in accordance with the editorial rules of the ISO/IEC Directives, Part 2 (see [www.iso.org/directives](http://www.iso.org/directives)).

Attention is drawn to the possibility that some of the elements of this document may be the subject of patent rights. ISO shall not be held responsible for identifying any or all such patent rights. Details of any patent rights identified during the development of the document will be in the Introduction and/or on the ISO list of patent declarations received (see [www.iso.org/patents](http://www.iso.org/patents)).

Any trade name used in this document is information given for the convenience of users and does not constitute an endorsement.

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This document was prepared by Technical Committee ISO/TC 155, *Nickel and nickel alloys*.

This second edition cancels and replaces the first edition (ISO 7524:1985), which has been technically revised. The main changes compared with the previous edition are as follows:

- the scope has been limited to ferronickels only;
- the former Clauses 5 and 7 have been technically revised;
- the former Annexes A and C have been deleted;
- the precision data have been updated.

Any feedback or questions on this document should be directed to the user's national standards body. A complete listing of these bodies can be found at [www.iso.org/members.html](http://www.iso.org/members.html).

# Ferronickels — Determination of carbon content — Infrared absorption method after induction furnace combustion

## 1 Scope

This document specifies an infrared absorption method after combustion in an induction furnace for the determination of the carbon content in ferronickels in the range of 0,004 % to 2,5 %.

The method is applicable to normal production operations. It uses commercially available equipment, which is calibrated using steel and/or ferronickel certified reference materials (CRMs).

## 2 Normative references

There are no normative references in this document.

## 3 Terms and definitions

No terms and definitions are listed in this document.

ISO and IEC maintain terminological databases for use in standardization at the following addresses:

- ISO Online browsing platform: available at <https://www.iso.org/obp>
- IEC Electropedia: available at <http://www.electropedia.org/>

## 4 Principle

Combustion of a test portion in a high-frequency induction furnace at high temperature in a current of pure oxygen, and in the presence of accelerators and fluxes.

Transformation of carbon into carbon dioxide and/or carbon monoxide.

Measurement by infrared absorption of the carbon dioxide and/or carbon monoxide carried by the current of oxygen.

## 5 Reagents

During the analysis, use only reagents of recognized analytical grade.

### 5.1 Oxygen, high purity (mass fraction minimum 99,5 %).

An oxidation catalyst [copper(II) oxide or platinum] tube heated at 600 °C, followed by appropriate carbon dioxide and water absorbents, shall be used when the presence of organic contaminants is suspected in the oxygen.

### 5.2 Inert ceramic (attapulugus clay), impregnated with sodium hydroxide and having particle sizes from 0,7 mm to 1,2 mm for the absorption of carbon dioxide.

### 5.3 Magnesium perchlorate [Mg(ClO<sub>4</sub>)<sub>2</sub>], having particle size from 0,7 mm to 1,2 mm for the absorption of moisture.

**6.1.3** Maintain a glass-wool filter ([5.4](#)) between the furnace chamber and the analyser and change as necessary. The furnace chamber, pedestal post and filter trap should be cleaned frequently to remove oxide residues.

**6.1.4** The manufacturer may recommend setting the programming unit to give a pre-burn period before oxygen enters the furnace chamber. The test portion should be at a red heat during the pre-burn period. When oxygen is introduced for the combustion stage, the temperature increases substantially.

**6.1.5** The temperature reached during combustion depends on the furnace, and the type and quantity of metal in the crucible. A high temperature (>1 700 °C) is maintained after the test portion is fused so that the carbon dioxide can be completely removed from the furnace to the infrared analyser.

**6.1.6** The flow rate of oxygen can vary from one instrument to another but is usually about 2,0 l/min during the combustion period.

**6.1.7** After the equipment has been idle for a few hours or after cleaning the furnace chamber or filters, the instrument should be stabilized as described in [8.1](#).

NOTE Features of commercial equipment are given in [Annex A](#).

## **6.2 Ceramic crucibles and lids.**

Ceramic crucibles, containing the sample and any additions that can be necessary, are required for the combustion. They shall be of precise dimensions for the system and fit the supporting pedestal post so that the test portion in the crucible is positioned correctly within the induction coil for heating.

Typical dimensions of combustion crucibles are: a height of 25 mm, an external diameter of 25 mm, an internal diameter of 20 mm, a wall thickness of 2,5 mm and a thickness of base of 8 mm. The dimension of the hole of the lid should be larger than 10 mm.

Crucibles and lids shall be as specified by the manufacturer of the instrumentation used and shall be capable of withstanding combustion in an induction furnace without evolving carbon-containing chemicals so that achieving and maintaining blank values within specification is possible.

In order to remove any carbon contamination, for the determination of carbon levels of less than 0,05 % (mass fraction), pre-ignite the crucibles in air or oxygen in a furnace for not less than 1 h at 1 100 °C and store in a desiccator or a closed container. A resistance furnace with a combustion tube through which a flow of oxygen passes may also be used.

Crucible lids, used to help retain the solid oxidation products in the hot zone of the induction coil, shall be pre-ignited in a similar manner.

**6.3 Crucible tongs**, capable of handling recommended crucibles ([6.2](#)).

## **7 Sampling and sample preparation**

**7.1** Sampling and sample preparation shall be carried out by normal agreed procedures or, in cases of dispute, by appropriate national standards.

**7.2** The laboratory sample is normally in the form of a powder, granules, millings or drillings and no further preparation of the sample is necessary.

**7.3** If it is suspected that the laboratory sample is contaminated with oil or grease from the milling or drilling process, it shall be cleaned by washing with high purity acetone and drying in air.

7.4 If the laboratory sample contains particles or pieces of widely varying sizes, the test portion should be obtained by riffing.

## 8 Procedure

**WARNING** — The risks related to combustion analysis are mainly burns in pre-igniting the ceramic crucibles and in the fusions. Use crucible tongs at all times and appropriate containers for the used crucibles. Normal precautions for handling oxygen cylinders shall be taken. Oxygen from the combustion process shall be removed effectively from the apparatus since a high concentration of oxygen in a confined space can present a fire hazard.

### 8.1 Preparing and stabilizing the instrument

8.1.1 Assemble the apparatus and prepare it for operation according to the manufacturer's instructions. Test the furnace and analyser to ensure the absence of leaks.

8.1.2 Condition and stabilize the equipment by combusting several samples, similar to those to be analysed, using appropriate fluxes and accelerators, before attempting to calibrate the system or determine the blank.

NOTE It is not necessary to use pre-ignited crucibles.

8.1.3 Allow the instrument to cycle several times with oxygen flowing and adjust the zero of the instrument.

### 8.2 Blank test and zero adjustment

#### 8.2.1 Carbon contents $\leq 0,1$ %

For each instrument range, transfer the selected mass of flux (5.5.2), to the nearest 0,005 g, into a pre-ignited crucible (6.2), add the selected mass of a CRM (5.6) having a very low carbon content, and then cover it with the selected mass of accelerator (5.5.3).

The type and mass of the flux and accelerator used in the blank test should accord with those used for determinations (details as in 8.4.1). Record the mass of the CRM. Place the crucible and contents on the furnace pedestal and operate the furnace in accordance with the manufacturer's instructions. Repeat the determination three more times. Average the results.

Subtract the carbon content of the CRM from the average to determine the blank value. If the blank is greater than 0,001 % (mass fraction) and the related standard deviation is greater than 0,000 2 % (mass fraction), find the cause of the problem, fix it and repeat the experiment.

Record the average blank value into the analyser in accordance with the manufacturer's instructions.

NOTE 1 The reading obtained corresponds to the blank due to the crucible, flux and accelerator.

If the analyser does not have automatic blank correction, the blank value should be subtracted from the total result prior to any calculation.

NOTE 2 An alternative procedure is to record the reading of the blank test and make the correction using a calibration graph.

#### 8.2.2 Carbon contents $> 0,1$ %

It is preferable not to carry out a blank correction. However, a check of the blank level shall be done. The value shall remain below 0,001 % (mass fraction).

## 8.3 Calibration

**8.3.1** Select a set of steel and/or ferronickel CRMs (5.6) for calibration and verification which, at a minimum, fall at the bottom, top and quartile points of each operational operating range.

**8.3.2** If the instrument has more than one carbon detector (measurement system), carry out the adjustment described in this section on each one.

Establish all experimental parameters for each range of carbon. Parameters to be specified include:

- crucible: to be pre-burned or not;
- flux and accelerator: type and mass;
- test portion: mass.

**8.3.3** For each detector (see 8.3.2) weigh an appropriate amount (usually 0,50 g) of a CRM having a carbon content corresponding to the top of each operating range into a pre-ignited crucible (6.2). Add the pre-selected amounts of flux (5.5.2) and accelerator (5.5.3) and combust as described in 8.4.1. Repeat this process twice.

If the results are situated in the interval “Certified Value  $\pm 2\sigma$ ”, adjust the net instrument reading to correspond to the certified content of carbon in the reference material, according to the manufacturer’s instructions.

**8.3.4** Check the linearity of the calibration by analysing at least three times a CRM having a carbon content in the middle of the range of the detector. The result shall be situated in the interval “Certified Value  $\pm 2\sigma$ ”.

Correct any non-compliant conditions before continuing to the next step.

## 8.4 Determination

**8.4.1** Weigh, to the nearest 0,001 g, 0,5 g to 0,6 g of the test sample, and transfer into a pre-ignited crucible (6.2) containing an appropriate amount of the selected flux (5.5.2), if required. Then cover the test sample with the appropriate quantity of accelerator (5.5.3) and, if necessary, place the lid on the crucible.

NOTE The flux and accelerator used will depend on the individual characteristics of the equipment and the type of material being analysed. For ferronickels, 1,0 g to 1,5 g of tungsten plus tin (7 to 9 parts of tungsten + 1 part of tin) can be used.

**8.4.2** Place the crucible and contents on the pedestal post of the furnace, raise to the combustion position and lock the system. Operate the furnace in accordance with the manufacturer’s instructions.

**8.4.3** Record the analyser reading and repeat the determination at least once.

NOTE 1 It is important that a high temperature be maintained after the sample is fused to ensure the complete transfer of the carbon dioxide from the furnace to the infrared analyser.

NOTE 2 Quiescent combustion is necessary to avoid splashing on to the crucible lid.

**8.4.4** Analyse, at least twice, CRMs having carbon contents slightly above and slightly below the content found for each unknown sample. The corresponding result shall be situated in the interval “Certified Value  $\pm 2\sigma$ ”. The CRMs used for checking the accuracy of the determination shall be different from those used for the calibration (see 8.3).

**8.4.5** If abnormal deviations appear, reject the results and restart the procedures described in [8.3](#) and [8.4](#).

## 9 Expression of results

### 9.1 Calculation

**9.1.1** If the instrument has been calibrated to give a read-out directly as a percentage by mass of carbon with automatic compensation for the mass of the test portion, take the average of the set of determinations carried out and report the result.

**9.1.2** If the instrument has been calibrated based on a 0,50 g test portion and does not have automatic mass compensation, divide each reading by the respective mass, expressed in grams, of the test portion. Average the determinations carried out and report the mean result.

**9.1.3** With some instruments it will be necessary to prepare a calibration graph of instrument reading versus the mass, in micrograms, of carbon. Read off the graph the mass, expressed in micrograms, of carbon in the test portion, correct for the blank and mass of the test portion. Average the determinations carried out and report the result.

### 9.2 Precision

Fourteen laboratories in four countries participated in an inter-laboratory validation test programme under the auspices of ISO/TC 155 involving three determinations of carbon at seven content levels (samples).

Each laboratory carried out two determinations under repeatability conditions as defined in [ISO 5725-1](#), i.e. one operator, same apparatus, identical operating conditions, same calibration and a minimum period of time. The third determination was carried out at a different time using the same apparatus with a different calibration.

The compositions of the samples used are given in [Annex B](#).

The results obtained were statistically evaluated in accordance with [ISO 5725-2](#) and [ISO 5725-3](#). They are reported in [Table 1](#).

The logarithmic relationships between the carbon content ( $m$ ) and the precision parameters ( $r$ ,  $R_w$  and  $R$ ), together with the corresponding correlation coefficients are:

$$\lg r = 0,653 \lg m - 1,932 \quad [\text{correlation coefficient} = 0,971]$$

$$\lg R_w = 0,761 \lg m - 1,582 \quad [\text{correlation coefficient} = 0,980]$$

$$\lg R = 0,638 \lg m - 1,383 \quad [\text{correlation coefficient} = 0,964]$$

The corresponding graphical representation is shown in [Annex C](#).

The smoothed values of the repeatability limit ( $r$ ) and reproducibility limits ( $R_w$  and  $R$ ) of the test results are summarized in [Table 2](#).

Table 1 — Results obtained from the precision test

Sample (%)	Ferronickel 1	Ferronickel 2	Ferronickel 3	Ferronickel 4	Ferronickel 5	Ferronickel 6	Ferronickel 7
Mean	2,496	1,033	1,460 4	0,475 5	0,096 1	0,036 3	0,004 24
$\sigma (r)$	0,008	0,004	0,008	0,002 8	0,000 5	0,000 4	0,000 17
$\sigma (R_w)$	0,013	0,012	0,018	0,006 1	0,001 3	0,000 4	0,000 21
$\sigma (R)$	0,025	0,021	0,025	0,007 2	0,003 0	0,000 9	0,000 75
$r$	0,022	0,011	0,023	0,007 8	0,001 3	0,001 2	0,000 48
$R_w$	0,036	0,034	0,048	0,017 0	0,003 7	0,001 3	0,000 57
$R$	0,069	0,059	0,070	0,020 2	0,008 3	0,002 5	0,002 10
CV (R)	0,98	2,03	1,71	1,52	3,07	2,49	17,7
Aim CV (R)	1,08	1,46	1,30	1,91	3,33	4,66	9,8
Max CV (R)	2,36	3,21	2,85	4,20	7,31	10,2	21,6
Assigned value	2,50	1,02	1,45	0,474	0,096	0,036	0,004 4

**Table 2 — Smoothed values of the repeatability and reproducibility limits**

<b>Carbon content</b>	<i>r</i>	<i>R<sub>w</sub></i>	<i>R</i>
0,004	0,000 3 <sub>2</sub>	0,000 3 <sub>9</sub>	0,001 2 <sub>2</sub>
0,005	0,000 3 <sub>7</sub>	0,000 4 <sub>7</sub>	0,001 4 <sub>1</sub>
0,01	0,000 6	0,000 8	0,002 2
0,02	0,000 9	0,001 3	0,003 4
0,05	0,001 7	0,002 7	0,006 1
0,1	0,002 6	0,004 5	0,009 5
0,2	0,004 1	0,007 7	0,014 8
0,5	0,007 4	0,015 4	0,026 6
1,0	0,012	0,026	0,041
2,0	0,018	0,044	0,064
2,5	0,021	0,053	0,074

## 10 Test report

The test report shall include the following information:

- a) all information necessary for the identification of the sample and the laboratory;
- b) the method used by reference to this document, i.e. ISO 7524;
- c) the results and the unit in which they are expressed;
- d) any unusual features noted during the determination;
- e) any operation not specified in this document, or any optional operation which may have influenced the results;
- f) signature of the responsible person;
- g) the date of the test and/or date of preparation or signature of the test report.

## Annex A (informative)

### Features of commercial high frequency induction furnaces and infrared carbon analysers

#### A.1 Combustion furnace

**A.1.1** The combustion furnace consists of an induction coil and a high frequency generator. The furnace chamber consists of a silica tube that fits inside the induction coil. This tube has metal plates at the top and bottom which are sealed to the tube by O-rings. Gas inlet and outlet points are made through the metal plates.

**A.1.2** The generator is usually a 1,5 kV.A to 2,5 kV.A, apparent power unit but the frequency used by various manufacturers may be different. Values of 2 MHz to 6 MHz, 15 MHz and 20,0 MHz have been used. The power from the generator is fed to the induction coil, which surrounds the silica furnace tube and is usually forced air cooled.

**A.1.3** The crucible containing the sample, flux and accelerator is supported on a pedestal post, which is precisely positioned so that, when raised, the metal in the crucible is correctly placed within the induction coil for effective coupling when the power is applied.

**A.1.4** The induction coil diameter, the number of turns and the furnace chamber geometry, and the power of the generator define the degree of coupling that can occur. These factors are defined by the instrument manufacturer.

**A.1.5** The temperature attained during combustion depends in part on the factors given in [A.1.4](#) but also on the characteristics of the metal in the crucible, the form of the test portion and the mass of material. Some of these factors may be varied to some extent by the operator.

#### A.2 Infrared gas analyser

**A.2.1** In most instruments, the gaseous products of combustion are transferred in a continuous flow of oxygen to the analyser system. The gases flow through an infrared cell, for example of the Luft type, where the absorbance of the infrared radiation due to carbon dioxide is measured and integrated over a pre-programmed time period. The signal is amplified and converted to a digital display of the percentage concentration of carbon.

**A.2.2** In some analysers, the products of combustion may be collected in oxygen in a fixed volume at controlled pressure and the mixture analysed for carbon dioxide.

**A.2.3** Electronic controls are usually provided for adjusting the zero of the instrument, compensating for the blank, adjusting the slope of the calibration line and correcting for the nonlinear response. The analyser generally has means of entering the mass of the samples for an automatic correction of the read-out. Instruments may also be equipped with an integrated automatic balance for weighing the crucibles, weighing the samples and transferring the mass to the analyser.

## Annex B (informative)

### Composition of the samples used for the validation precision test

The compositions of the samples used for the validation precision test are listed in [Table B.1](#).

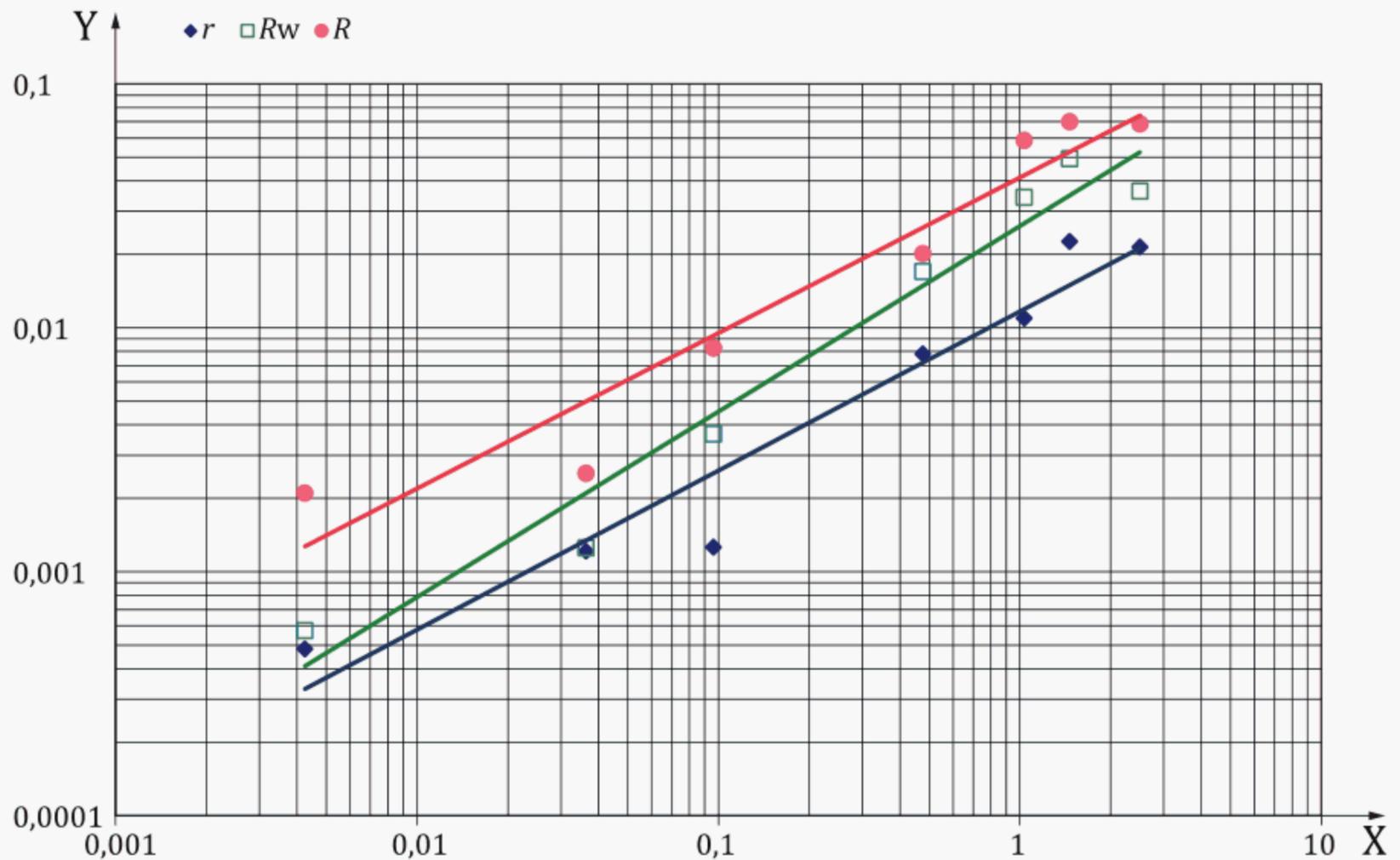
**Table B.1 — Compositions of the samples used for the validation precision test**

Sample no.	C	Si	P	S	Cr	Ni	Co	Cu	Fe
Ferronickel 1	2,50	1,04	0,020	0,276	1,94	14,96	0,98	0,117	77,85
Ferronickel 2	1,02	0,39	0,032 5	0,113	1,49	18,63	0,60	0,049	77,66
Ferronickel 3	1,45	0,22	0,011	0,049	1,08	25,95	0,46	0,015	70,74
Ferronickel 4	0,474	1,64	0,006 0	0,013	0,72	30,77	1,27	0,140	64,66
Ferronickel 5	0,096	3,33	0,009 2	0,009 5	0,36	40,54	1,87	0,191	52,77
Ferronickel 6	0,036	4,49	0,004 2	0,002 0	0,133	49,07	2,01	0,250	42,74
Ferronickel 7	0,0044	0,25	0,021	0,04	0,46	33,26	0,45	0,072	

## Annex C (informative)

### Graphical representation of the precision data

Figure C.1 shows the logarithmic relationships between the carbon content ( $m$ ) and the corresponding repeatability ( $r$ ) and reproducibility parameters ( $R_w$  and  $R$ ).



**Key**

X carbon content, in % (mass fraction)

Y precision, in %

**Figure C.1 — Logarithmic relationships between the carbon content ( $m$ ) and the corresponding repeatability ( $r$ ) and reproducibility parameters ( $R_w$  and  $R$ )**

## Bibliography

- [1] [ISO 5725-1](#), *Application of statistics — Accuracy (trueness and precision) of measurement methods and results — Part 1: General principles and definitions*
- [2] [ISO 5725-2](#), *Accuracy (trueness and precision) of measurement methods and results — Part 2: Basic method for the determination of repeatability and reproducibility of a standard measurement method*
- [3] [ISO 5725-3](#), *Accuracy (trueness and precision) of measurement methods and results — Part 3: Intermediate measures of the precision of a standard measurement method*
- [4] [ISO 9556](#), *Steel and iron — Determination of total carbon content — Infrared absorption method after combustion in an induction furnace*

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